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Ariane Sousa-Batista, Natalia Arruda-Costa, Bartira Rossi-Bergmann, Maria Inês Ré. Improved drug loading via spray drying of a chalcone implant for local treatment of cutaneous leishmaniasis. Drug Development and Industrial Pharmacy, Taylor & Francis, 2018, 44 (9), pp.1473-1480. 10.1080/03639045.2018.1461903. hal-01984692

HAL Id: hal-01984692 https://hal-mines-albi.archives-ouvertes.fr/hal-01984692

Submitted on 7 Nov 2019

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Improved drug loading via spray drying of a chalcone implant for local treatment of cutaneous leishmaniasis

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ABSTRACT

Current chemotherapy of cutaneous leishmaniasis (CL), even the mildest forms, encompasses multiple and painful injections with toxic drugs that cause systemic adverse effects. Recently, we showed the promising use of poly(lactic-co-glycolic acid) (PLGA) microparticles loaded with an antileishmanial nitrosylated chalcone (CH8) for effective, safe, local, and single-dose treatment of CL. Here, we proposed to optimize the delivery system by increasing the CH8 loading in PLGA-microparticles using spray drying instead of emulsification-solvent evaporation. The effect of solvent composition and polymeric matrix changes on thermal properties, loading efficiency, particle size, morphology, and spatial drug distribution of the CH8-loaded microparticles was evaluated. The results showed that spray drying allowed a higher CH8 content (18% w/w), as contrasting with the previous solvent evaporation technique that maximally incorporated 7.8% of CH8. *In vitro* studies on 96-hour incubation with *L. amazonensis*-infected macrophages showed that entrapment in spray-dried PLGA microparticles rendered CH8 safer, preserved its antileishmanial activity, and did not affect its antioxidant properties.

KEYWORDS

Cutaneous leishmaniasis; leishmania; chemotherapy; chalcone; implant; polymeric microparticle; spray drying

Introduction

Leishmaniasis is a complex of disease caused by the infection with the protozoan parasites of the genus *Leishmania* that are transmitted by the bite of phlebotomine sandflies. Cutaneous leishmaniasis (CL) is a dermal disorder witch cause high morbidity levels with a wide spectrum of clinical manifestations [1,2]. This disease is widespread in the world with approximately 0.7 to 1.2 million cases each year [3]. The first choice for CL treatment is the pentavalent antimonial drugs used since 1960's. Pentamidine, amphotericin B, and paromomycin are considered as second choice drugs. However, all these available drugs can present hazardous side effects, emergence of drug-resistant, and variable efficacy in different geographic areas and *Leishmania* species [1,4]. The search for new drugs is urgently needed.

Different studies have been appointed chalcones and their derivatives as potential drug candidates for leishmaniasis treatment [5–9]. A new chalcone derivate, 3-nitro-2'-hydro-4',6'-dimethoxychalcone (CH8) synthesized and patented by our group has shown a promising activity against both promastigotes and intracellular amastigotes of *Leishmania amazonensis*. In a murine model of cutaneous leishmaniasis the CH8 displayed the same efficacy of the reference drug (Pentostan[®]) in a dose 10 times lower [6].

To be successful, the treatment of CL should be effective with low doses, local effect, and safe [10]. Intralesional antimonial administration has been explored and demonstrated a good effect, although this treatment is based on multiple injections that induce local pain and may cause side effects, like anaphylactic shock [11,12]. Topical treatments are an attractive alternative due to easier administration, lower side effects, and patient-compliance. Nonetheless most of the studies using topical formulations with paromomycin and amphotericin B presented low efficacy, due to their poor permeability through the skin, which need the use of permeation enhancers that promote an irritant effect and hamper their approval [10,13–15]. Another interesting alternative for the local treatment of CL is the use of drug delivery systems. The polymeric implants systems have the advantages such as local action, reduced side effects, and improved patient compliance [16]. In leishmaniasis the implants based on microparticles can be up taken by macrophages, the main cell infected, which potentiate the drug effect [17]. Poly(lacticco-glycolic acid) (PLGA) implants are extensively applied in the clinical field due to their biocompatibility and biodegradability. Fifteen PLGA implants have been already approved by FDA from which, 11 are based on microparticles with applications in different diseases [18]. However, PLGA microparticle implants are poorly explored for the local treatment of CL.

Recently our group produced a PLGA microparticle implant loaded with 7.8% of CH8 prepared by emulsion-solvent evaporation method. This formulation administrated in a single dose by intralesional route demonstrated the same efficacy than the free drug in three doses in mice infected with *L. amazonensis* [19]. However, the emulsion-solvent evaporation method has many process variables that can hinder the scale up. In addition, we believe that the increase of the drug load in the microparticle implant can improve its effectiveness in a single dose and reduce the treatment cost with lowers amounts of polymer per dose. The goal of this study was to increase the CH8 load in implant based on PLGA microparticles generated by spray drying as an alternative production method. The effect of some parameters such as the composition of the polymeric matrix and the type of solvent was studied in the present work.

Materials and methods

Materials

The following chemicals were used as received: PLGA 5004 (lactide–glycolide molar ratio of 50:50, inherent viscosity of 0.41 dl/g)

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was purchased from PURAC Biomaterials (Gorinchem, The Netherlands); polyvinylpyrrolidone K17 (PVP K17) (Kollidon_17 PF endotoxinfree, BASF AG, Ludwigshafen, Germany); dichloromethane (DCM - Sigma-Aldrich, St. Louis, MO); ethyl acetate (EA - Fluka); acetonitrile (Sigma-Aldrich); phosphoric acid (FlukaTM, Seelze, Germany). The synthetic Chalcone 3-nitro-2-hidroxi-4,6-dimetoxichalcona was prepared as previously shown and patented by our group [6,20].

Solubility parameter estimation

The Hansen solubility parameter (δ) [21] was used to estimate the affinity between CH8 and PLGA, and PVP polymers and improved understanding of polymer effects in the final solid state of spraydried microparticles. The Hansen solubility parameter was calculated from the chemical structures of CH8, PLGA, and PVP using the Hoftyzer and Van Krevelen group contribution method [22] according to Equation (1):

$$\delta = \left(\delta_d^2 + \delta_p^2 + \delta_h^2\right)^{1/2} \tag{1}$$

where:

ð

$$\delta_d = \frac{\sum_i Fdi}{V}; \ \delta_p = \frac{\sqrt{\sum_i F^2 pi}}{V}; \text{ and } \delta_h = \sqrt{\frac{\sum_i Ehi}{V}}$$

V is the molar volume and *Fdi*, *Fpi*, and *Ehi* are the group contributions for the different types of interactions (dispersion forces, polar interactions, and hydrogen bonding, respectively), which are reported in the literature [21].

Preparation of CH8-loaded polymeric microparticles

CH8-loaded polymeric microparticles were prepared using a Buchi B-290 spray-dryer (Buchi Labortechnik AG, Flawil, Switzerland) equipped with lnert Loop B-295 and an integrated two-fluid 2 mm nozzle. Compressed nitrogen dispersed the liquid into fine drop-lets, which were consequently dried in the drying chamber and deposed in the cyclone. Drying conditions are given as follows for all prepared samples: aspirator 85%; pump flow rate 360 ml/h and compressed nitrogen flow rate 500 l/h. The inlet temperature was set to 49 ± 1 °C and the resultant outlet temperature was 38 ± 2 °C. The experiments were made in duplicate.

The drug:polymer mass proportion (dry basis) was fixed at 1:5. PLGA alone or a mixture of PVP K17 and PLGA of 1:10 were used. The feeding solution was prepared by dissolving CH8 in a 1.5% (w/v) solution of PLGA or PLGA/PVP K17 in dichloromethane or in a mixture of ethyl acetate in dichloromethane (1:2).

The individual polymers were also spray-dried from organic solutions. They were used for comparison purposes. All obtained powders were collected in glass containers and stored at -20 °C before further studies.

CH8-loaded polymeric microparticles were also produced by emulsification-solvent evaporation method. The preparation protocol was previously described by our group [19]. In our previous studies, a maximum CH8 loading of 7.8% was obtained by following this experimental protocol. The aim is to compare the performance of the CH8-loaded PLGA microparticles prepared by the two different processes, spray drying and emulsion-solvent evaporation.

Table 1 summarizes all studied formulations.

Physico-chemical characterization of the CH8-loaded polymeric microparticles

Thermal analyses

Thermal analyses were performed using a differential scanning calorimeter DSC Q2000 with the base module and modulate-DSC (TA Instruments, Leatherhead, UK). Samples were heated in non-hermetic aluminum pans at a rate of $5 \,^{\circ}$ C/min from 15 to 200 $^{\circ}$ C under nitrogen flow of 50 ml/min using an empty sealed pan as reference. DSC modulated technique (mDSC) was used to increase the resolution and the sensibility for weak transitions. In mDSC experiments, the samples were heated by a sinusoidal program of $2 \,^{\circ}$ C/min from 20 to 150 $^{\circ}$ C, with the modulation period of 40 s and amplitude 0.8 $^{\circ}$ C.

Drug loading

Drug loading was determined by dissolving 2 mg of PLGA or PLGA/PVP microparticles in 5 ml of acetonitrile and analyzing the sample using high-performance liquid chromatography (HPLC). HPLC was performed on an Agilent chromatographic system (1100, Agilent), and a Grace C18 column (250 \times 4.6 mm 5 μm – 201TP54, Grace, Munich, Germany). Acetonitrile: phosphoric acid 0.01% in water (80:20, v/v) was used as the mobile phase at a flow rate of 1.0 ml/min. The ultraviolet detection was conducted at 337 nm. Data was analyzed using the ChemStation software CA). (Agilent, Palo Alto, The calibration curve was

Table 1. Formulation parameters and thermal analysis (DSC) of the polymeric particles prepared by spray drying and by emulsion-solvent evaporation methods.

Code formulation	Polymer	Solvent	Particles characteristics						
			Size (µm)					Thermal properties	
			D10	D50	D90	Span	Percentage CH8 loaded	<i>T</i> g (°C)	7m _{onset} (°C)
Spray drying meth	od								
Blank (polymeric)	particles								
PD	PLGA	DCM	1.93	10.19	22.71	2.04	_	45.4	-
PDE	PLGA	DCM:EA	1.46	6.21	16.63	2.44	-	46.3	-
VD	PVP	DCM	-	-	-	-	-	145.5	-
PVD	PLGA:PVP	DCM	2.13	5.30	14.47	2.33	_	45.9	-
PVDE	PLGA:PVP	DCM:EA	1.88	4.35	12.18	2.37	-	48.1	-
								154.0	
CH8-loaded (polyn	neric) particl	es							
CH8/PD	PLGA	DCM	2.69	12.34	25.82	1.87	18.34 ± 0.02	38.4	134.6
CH8/PVD	PLGA:PVP	DCM	1.58	5.75	12.82	1.96	17.82 ± 0.04	40.9	136.0
CH8/PDE	PLGA	DCM:EA	1.53	8.40	17.99	1.96	18.10 ± 0.01	38.5	141.7
CH8/PVDE	PLGA:PVP	DCM:EA	1.66	8.97	19.11	1.95	18.14 ± 0.02	38.5	139.2
Emulsion-solvent e CH8-loaded (polyn	vaporation neric) particle	method es							
CH8/PLGA	PLGA	DCM	0.60	5.49	13.40	2.30	7.84 ± 1.50	48.6	150.5

y = 51.75x + 410.6 (r = 0.9992, n = 3), which exhibited linearity over a concentration range of $50-300 \mu$ g/ml of CH8. The drug loading was calculated by the following Equation (2):

$$Drug \ loading(\%) = \frac{Weight \ of \ drug \ in \ spray - Dried \ microparticles}{Weight \ of \ spray - Dried \ microparticles} \times 100\%$$
 (2)

Size, morphology and drug distribution within the CH8-loaded polymeric microparticles

Particle size was determined by laser light scattering (Mastersizer 2000, Malvern, Worcestershire, UK) using a Fraunhofer diffraction model for the analysis of the raw data. The particles were dispersed in 0.5% (w/w) Tween-20 solution and subsequently treated with ultrasound for 20 s (20 kHz, 50 W) [23]. Particle diameter was expressed as the values of D90, D50, and D10, where 90th, 50th, and 10th percent of the population lies below this value. Polydispersity was given by span index, calculated by (D90 – D10)/D50.

The morphology and the surface structure were assayed by field emission gun scanning electron microscopy (FEG-SEM) using Philips XL30 ESEM-FEG instrument (Philips, Eindhoven, Netherlands) operating at an acceleration voltage of 20 kV under nitrogen atmosphere. The microparticles were fixed on an aluminum stub using double-sided carbon tape and was coated by sputtering with gold for four minutes at 18,000 mA using a Polaron SC7640 (Quorom Technology, England).

The drug distribution in the spray-dried microparticles was mapped by Raman microscopy (Alpha 300 R Raman-AFM spectrophotometer, WITEC GmbH, Germany). The spectra were obtained at room temperature and 50X magnification using a confocal laser wavelength of 532 nm (Nd:YAG laser) and ultrahigh-throughput (UHTS 300) spectroscopy system with a CCD charge-coupled device (CCD) as detector. Each spectral scan was collected for 0.5 s of integration using ten accumulations and each sample was analyzed three times.

Biological evaluation

Antileishmanial activity

The antileishmanial activity of CH8-loaded polymeric microparticles was evaluated against intracellular amastigote of *L. amazonensis* (Josefa strain). Adherent mouse peritoneal macrophages (5×10^5 cells) were plated on coverslips and infected with 5×10^6 promastigotes for 4 h at 34 °C, after which non-internalized parasites were removed with warm phosphate buffered saline (PBS). Then, to allow for intracellular amastigote growth, the cells were cultured for a further 24 h at 37 °C in Roswell Park Memorial Institute medium (RPMI) supplemented with 5% heat-inactivated fetal bovine serum (HIFBS, Cultilab, Brazil).

Next, the infected macrophages were treated for 96 h in triplicates with a range of concentrations (0.5, 5, and 50 μ g/ml) of free CH8 or CH8 encapsulated in microparticles produced by spray drying (CH8/ PDE and CH8/PVDE) or emulsification-solvent evaporation (CH8/ PLGA). At the end of treatment time, the cell monolayers were stained with Giemsa and the numbers of amastigotes in total 200 macrophages per coverslip were counted in Neubauer chambers. The results were expressed both as amastigotes per 100 macrophages and IC₅₀ (drug concentration that inhibited parasite growth by 50%).

Anti-macrophage cytotoxicity

The 'resazurin reduction test' is a very simple and versatile way to measure cell cytotoxicity. This method is based on the enzymatic reduction of resazurin sodium salt (blue and nonfluorescent) to resorufin (pink and highly fluorescent) by metabolically active cells [24]. For evaluation of the anti-macrophage cytotoxicity,

macrophages were infected and treated for 96 h with CH8 encapsulated or not, as described above for antileishmanial activity. Then, 10% (v/v) of resazurin solution (0.125 mg/ml;Sigma-Aldrich) was added in the medium and cells were cultured for additional four hours at 37 °C. Following incubation, the fluorescence was measured at 555/585 nm (Bio-Tek Powerwave XS, Winooski, VT). The results were reported as the percentage of viable cells normalized to untreated cells. The 50% cytotoxic concentration (CC₅₀) values were determined by logarithmic regression analysis.

Nitric oxide (NO) production

The NO induction by microparticles was assessed in both uninfected and *L. amazonensis*-infected macrophages, as described for antileishmanial activity. After 24 h of infection, the cells were treated for additionally 96 h with CH8 encapsulated or not (50 μ g/ml) or equivalent amounts of blank microparticles. As controls untreated cells and 1 μ g/ml of lipopolysaccharide (LPS *Escherichia coli*, Sigma-Aldrich)-treated cells were used. The stable sodium nitrite (NaNO₂) product was measured in the culture supernatants at 570 nm (BIO-TEK, Powerwave XS) by Griess colorimetric method and NaNO₂ standard curve [25].

Statistical analysis

Unpaired t test with Welch's correction was performed to demonstrate statistical differences (p < .05) between two groups. IC₅₀ values were calculated from a curve of sigmoidal dose-response. All analysis were conducted using GraphPad Prism 5 software (GraphPad Software, La Jolla, CA).

Results and discussion

Physicochemical characterization of CH8-loaded polymeric microparticles

Thermal properties

All heating curves obtained from DSC analyses are grouped in Figure 1, including the heating curve of a physical mixture between CH8 and PLGA in a mass proportion of 1:2. The thermal profile of unprocessed CH8 crystals shows a distinct melting endotherm (Tm_{onset}) at 166.3 °C with a ΔH_m of 89.21 J/g. Amorphous PLGA has a glass temperature (Tg) in the range of 45.4–46.3 °C (PD and PDE, respectively), whereas spray-dried PVP has a Tg of 145.5 °C (VD).

The fast solvent evaporation of solvents from organic solutions containing two dissolved constituents, drug and polymer, in most cases leads to the formation of amorphous solids dispersions during spray drying [26]. An amorphous solid dispersion constituted of CH8 and PLGA or PLGA/PVP could be theoretically expected from the analysis of their molecular structure and solubility parameters (δ). The solubility parameters (δ) for CH8 from its molecular division in groups was calculated to be 21.9 MPa^{1/2}. The δ t calculated values for PLGA (24.36 MPa^{1/2}) and PVP (23.36 MPa^{1/2}) are close to data found in the literature [21,27,28]. It is generally believed that favorable interactions and a uniform phase will result when the difference in δt values between two components ($\Delta \delta t$) is less than 7 MPa^{1/2}, while unfavorable interactions and phase separation will result when $\Delta \delta t$ > 10 MPa^{1/2} [29]. In the present case, $\Delta\delta$ t between CH8 and all polymers used is lower than 3 MPa^{1/2} and CH8 is likely to be miscible with these polymers in the spray-dried microparticles.

However, despite the favorable outcome in the $\Delta \delta t$, the spraydried CH8-loaded polymeric microparticles are not constituted by a homogeneous mixture of amorphous components. In fact, from the DSC results (Table 1), all CH8 loaded-polymer particles present a clear *T*g corresponding to the *T*g of the polymers and in addition a melting endotherm near to the melting point (*T*m) of the CH8 crystalline form. The presence of the polymers in the spraydried microparticles led to a depression in the melting point of the drug in the binary mixtures. This finding could be indicative of mixing between two components, as already found and discussed in the literature for other binary mixtures of drug and polymer [30]. In theory, melting of a crystal occurs at the temperature when the chemical potential of the crystal is equal to the chemical potential of the melt. Addition of an amorphous polymer to the crystal (if miscible) reduce the chemical potential of the crystalline material leading to the reduction of its melting temperature [31].

DCM alone and a mixture of DCM:EA (1:2) were used as solvents for drug and polymers. The influence of the initial solvent composition on the thermal properties of the spray-dried microparticles can be observed with a lower depression of *T*m when DCM was used as solvent (Table 1). PLGA is less soluble in EA than in DCM, probably the lower solubility of PLGA in the mixtures DCM:EA led to a more rapid precipitation of the polymer interfering in drug-polymer interaction.

Another investigation carried out in this study was the introduction of PVP (10% w/w) in the polymeric matrix. PVP is a biocompatible and hydrophilic polymer, capable of promoting the formation of solid amorphous dispersions by forming electrostatic interactions with several drugs [32]. Irrelevant differences were observed in the *T*m of the drug in spray-dried PLGA microparticles containing PVP in the polymeric matrix compared to the CH8-loaded PLGA microparticles without PVP (Table 1).

Drug loading efficiency

The spray-dried CH8-loaded polymeric microparticles covered an actual drug loading range of 17.8 to 18.3% w/w, compared to 7.8% w/w for the microparticles obtained by emulsion-solvent evaporation (Table 1). The changes in the polymeric matrix or solvent composition did not significantly modify the high drug loading efficiency (>96%) for the spray-dried microparticles.

Particle size, morphology, and spatial drug distribution

Despite of all modifications in the spray-dried formulations, the particle size ranged from 1.5 (D10) to $26 \,\mu$ m (D90; Table 1), in a very close way. In comparison, CH8-loaded PLGA microparticles prepared by emulsion and solvent evaporation showed a smaller size range (from 0.6 (D10) to 13.4 μ m (D90). This variety in the microparticle distribution size is interesting because the microparticles below 10 μ m can be efficiently uptake by macrophages [33], the main cell infected in leishmaniasis, while the largest were desired to obtain injectable microparticles for depot applications [34,35].

All particles were analyzed by FEG-SEM and showed a similar topography. Figure 2 depicts a representative picture of PLGA microparticles loaded with CH8 showing that they are spherical and



Figure 1. DSC thermograms. (A) Blank microparticles (PVDE; PVD; VD; PDE, and PD); (B) CH8's microparticles produced by spray drying (CH8/PVDE; CH8/PDE; CH8/PVD, and CH8/PD) and by emulsion-solvent evaporation (CH8/PLGA) methods, unprocessed CH8 crystals (CH8) and physical mixture between CH8 and PLGA was in a mass proportion of 1:2 (CH8 + PLGA).



Figure 2. SEM micrograph of PLGA microparticles with CH8. (A) Intact particles; (B) Particles rubbed out by friction, showing the inside with crystal shapped CH8 chalcone (white arrows).

have a relatively smooth surface (Figure 2(A)). The intentional wear in the surface of the particles revealed an interior probably rich in drug crystals (Figure 2(B)). To confirm this statement, the spatial distribution of components inside the microparticles was evaluated by RAMAN microscopy mapping. For this, surface $(20 \times 20 \,\mu\text{m})$ and deep spectra $(9 \times 9 \,\mu\text{m})$ have been taken for accuracy.

RAMAN microcopy has been used to study the distribution of drugs in drug delivery systems without the need to mark the molecule [36]. Our results confirmed that CH8 loaded-polymeric microparticles have an outer phase composed of a mixture of drug and polymer (peaks around $1600 \, \mathrm{cm}^{-1}$ of CH8 and peaks between 2800 and $3200 \, \mathrm{cm}^{-1}$ of PLGA or PLGA/PVP) and an inner phase where the drug predominates (Figure 3). All particles studied showed the same polymer and drug distributions, although it was not possible to differentiate the PLGA and PVP spatial distributions, because of their similar spectra with mean peaks between 2800 and $3200 \, \mathrm{cm}^{-1}$.

No changes were observed in the microparticles characteristics, even though the physical composition of the microparticles were altered. The microparticles produced with DCM:EA as solvent phase was chosen to continue the *in vitro* experiments, because they were produced with a lower ratio of DCM in the feed solution, creating a more sustainable method.

Biological evaluation of the CH8-loaded PLGA microparticles

Antileishmanial activity

All CH8 formulations were active against intracellular amastigotes of *L. amazonensis.* As demonstrated before, encapsulation of CH8

in PLGA microparticles preserved the CH8 antileishmanial activity [19] and the addition of PVP in the polymeric matrix did not modify this effect. Nonetheless, encapsulation of CH8 into microparticles slightly increased its IC_{50} from $0.4 \,\mu$ g/ml (free CH8) to $0.9-1.9 \,\mu$ g/ml (CH8 loaded polymeric microparticles), as shown in Figure 4. That is probably due to the slow drug release from microparticles in the intracellular milieu. The process that led to higher CH8/PLGA activity was emulsification-solvent evaporation ($IC_{50} = 0.9 \,\mu$ g/ml), as compared with spray drying (CH8/PVDE: $1.8 \,\mu$ g/mL and CH8/PDE: $1.9 \,\mu$ g/ml). Since CH8/PLGA



Figure 4. Antiamastigote activity. *L. amazonensis*-infected macrophages were treated with free or encapsulated CH8 (0; 0.5; 5, and 50 µg/ml) or left untreated for 96 h at 37 °C. The intracellular amastigote loads, in total 100 macrophages/ coverslip, were counted in Giemsa-stained cells. The half maximal inhibitory concentration (IC₅₀) values were determined by logarithmic regression analysis. Means ± SD (n = 3).



Figure 3. Representative Raman mapping of the microparticles with peaks of CH8 (peaks around 1600 cm^{-1}) and PLGA (peaks between 2800 and 3200 cm^{-1}) in the surface ($20 \times 20 \ \mu\text{m}$) and deep spectra ($9 \times 9 \ \mu\text{m}$).



Figure 5. Microparticles cytotoxicity. *L. amazonensis*-infected macrophages were treated for 96 h with free or encapsulated CH8 (0; 0.5; 5, and 50 μ g/ml) or blank microparticles (50 μ g/ml of encapsulated CH8-equivalent doses), or left untreated. Following treatment, the cell viability was evaluated after cell incubation for 4 h at 37 °C with 10% of resazurin (0.125 mg/ml) and fluorescence measurement at 555/585 nm. The result was plotted as percentage cell viability considering untreated as 100% of cell viability. Untreated fluorescence units = $2.07 \times 10^7 \pm 1.58 \times 10^5$. Mean \pm SD (n = 3). *p < .05 vs. CH8 group in the same concentration.



Figure 6. NO production. *L. amazonensis*-infected and uninfected peritoneal macrophages were treated with blank microparticles, encapsulated, or free CH8 (50 μ g CH8/ml and equivalent amounts of particle), or left untreated. Treatment with LPS (1 μ g/ml) was used as a positive control. After 96 h at 37 °C, culture supernatants were collected to assess the production of NO by colorimetric assay using Griess reagent. Mean ± SD (n = 3).

microparticles can be efficiently internalized by macrophages [19], the higher CH8/PLGA activity may be related to its higher uptake by macrophages due to the smaller size.

Anti-macrophage cytotoxicity

PLGA microparticles are widely used in different FDA-approved products, supporting their safety [18,37]. CH8 safety and antileishmanial activity have also been described [6]. However, in this work, the safety of their formulation was evaluated. Despite of the high antileishmanial activity of CH8, cytotoxicity was observed in higher concentrations of these drug [19]. Our *in vitro* studies on 96-hour incubation with *L. amazonensis*-infected macrophages showed that all CH8-loaded microparticles tested (CH8/PLGA, CH8/PDE, and CH8/PVDE) displayed higher safety than free CH8 (CC₅₀ >50 and 36.8 µg/ml, respectively; Figure 5). The slight decrease in cell viability observed in CH8-loaded polymeric microparticles may be a drug effect, since blank microparticles (with or without PVP) did not modify cell viability.

NO production

Macrophage microbicidal mechanisms such as reactive oxygen species (ROS) and NO production play an important role in the control of *Leishmania* infection [38,39]. It is also known that NO induction directly correlates with decreased parasite loads in patients with cutaneous leishmaniasis [40]. Additionally, endocytosis itself may induce physical changes in the plasma membrane, which culminates with the activation of nitric oxide synthase and NO generation [17]. Then, to verify if the microparticle uptake interfered with oxidative mechanisms, NO production was evaluated in infected and uninfected macrophages following addition of free or encapsulated CH8 to cells. However, no NO production was induced (Figure 6) and was also seen to be compatible with the antioxidant [41] and anti-inflammatory properties of chalcones, even with a decrease in NO production [42,43]. In all, these results show that PLGA microparticles do not revert the antioxidative mechanisms of CH8 in macrophages. So, the CH8-microparticles antileishmanial activity was due to the direct action of the intracellularly released CH8 and was not to due to the macrophage activation, compatible with the direct action of CH8 in the *Leishmania* mitochondria [5,8].

Conclusion

Alternative treatments of CL using safer drugs and reduced doses are urgently need. Polymeric implants emerge as an attractive strategy to solve this problem. However, to the best of our knowledge, this is the first attempt to develop chalcone implant based on spray-dried PLGA microparticles. In the current work, spray drying allowed the efficiently production of CH8 loaded-PLGA microparticles with an increase around 125% in the CH8 chalcone content in relation to the microparticles obtained previously by an emulsionsolvent evaporation process. Besides, in our *in vitro* studies, the CH8-microparticles showed to be safe and active against *Leishmania amazonensis* amastigotes. In future studies, we propose to evaluate *in vivo* the efficacy and safety of encapsulated CH8 in experimental models of cutaneous leishmaniasis.

Acknowledgements

The authors acknowledge to S. Delconfetto, C. Rolland, L. Haurie, and P. Accart from Rapsodee Centre for DSC, SEM, Raman and particle size measurements help and analyses, respectively. The first author also acknowledges CNPq (Brazilian Research Council for Scientific and

Technological Development) and CAPES (Brazilian Federal Agency for Support and Evaluation of Graduate Education within the Ministry of Education of Brazil) for providing PhD Research Fellowship.

Disclosure statement

No potential conflict of interest was reported by the authors.

Funding

This work was supported by funding from Brazilian Research Council for Scientific and Technological Development (CNPq) [grant number 402787/2013-7].

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